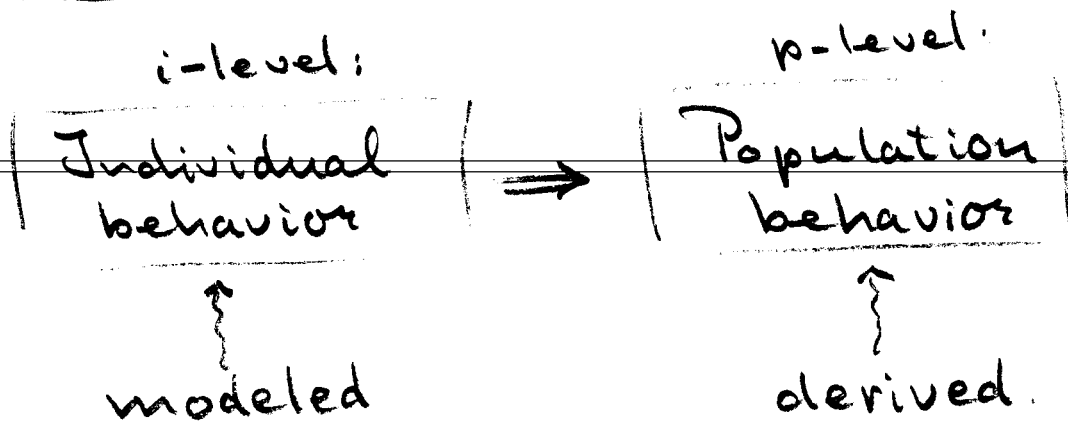


Previous lecture



Basic processes on i-level

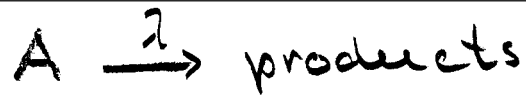
- Interactions & reactions
- Growth & development
- movement in space.

Interactions & reactions

- Monomolecular reaction (MR)
- Bimolecular reaction (BR)
- Reaction networks.

Monomolecular reaction

i-level: A : single individual.



(Poisson-process with rate λ)

p-level: a : population density

$$\frac{da}{dt} = -\lambda a$$

(Law of the large numbers)

Biological examples

Cell division

Asexual reproduction

Maturation

Death

Recovery from illness

... may all be modelled by monomolecular reactions.

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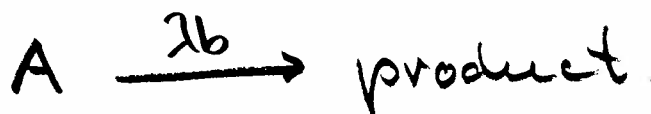
Today:

Bimolecular reaction



Two particles meet and react
For a given A particle, the number of meetings with any B particle per unit of time is proportional to the concentration b of B particles.

\Rightarrow (*) can also be represented by the reaction



(Poisson process with rate λb)

$$\Rightarrow \frac{da}{dt} = -(\lambda b)a = -\lambda ab$$

Likewise:

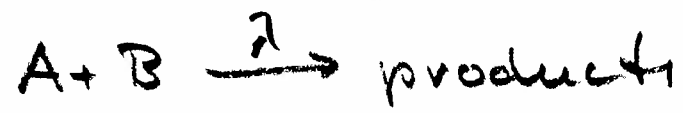
$$\frac{db}{dt} = -(\lambda a)b = -\lambda ab$$

Principle of mass-action

In a bimolecular reaction, the probability per unit of time that a given particle undergoes a reaction is proportional to the concentration of the kind of particles it is reacting with.

Bimolecular reaction

i-level A, B individuals of different kinds.



(assuming mass action)

p-level a, b population densities

$$\left\{ \begin{array}{l} \frac{da}{dt} = -\lambda ab \\ \frac{db}{dt} = -\lambda ab \end{array} \right.$$

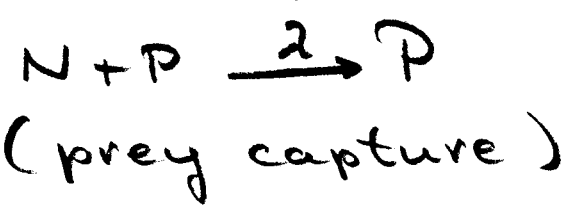
(Law of large numbers).

1st Example

| i-level |

P predator individual

N prey individual



| p-level |

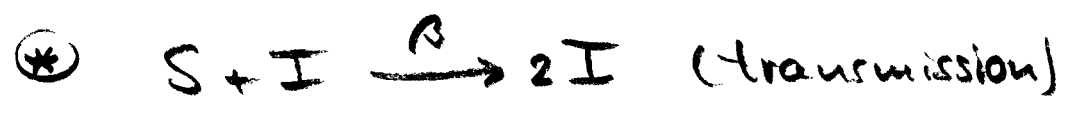
p, n pop. dens.

$$\left\{ \begin{array}{l} \frac{dp}{dt} = 0 \\ \frac{dn}{dt} = -2np \end{array} \right.$$

2nd Example

| i-level |

I: infected, S: susceptible



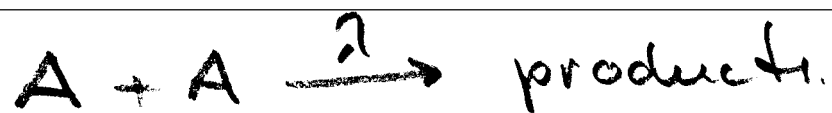
| p-level |

i, s pop dens.

$$\left\{ \begin{array}{l} \frac{ds}{dt} = -\beta si \\ \frac{di}{dt} = \underbrace{-\beta si}_{\text{left side of } (*)} + \underbrace{2\beta si}_{\text{right side of } (*)} = +\beta si \end{array} \right.$$

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Bimolec. reaction between
two identical particles:



Write as



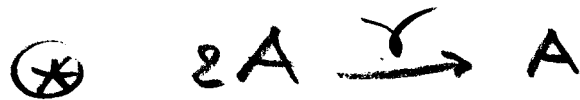
hence

$$\frac{da}{dt} = -(\lambda a) a = -\lambda a^2$$

(and not $\frac{da}{dt} = -2\lambda a^2$)

Example.

i-level | A : individual



(interference competition)

p-level | a : pop. dens.

$$\frac{da}{dt} = -\lambda a^2 + \frac{1}{2} \lambda a^2 = -\frac{1}{2} \lambda a^2$$

↑
left side
of $\textcircled{*}$.

right side of $\textcircled{*}$:

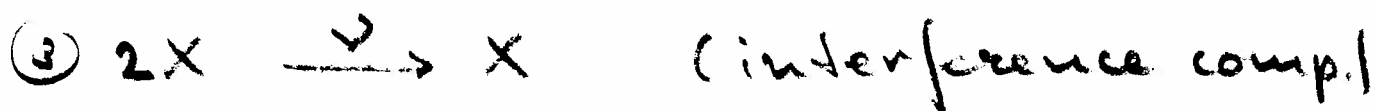
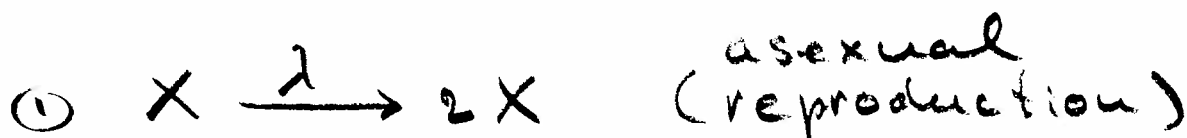
(A re-appears at
half the rate
as A disappears
on the left)

Reaction networks.

i-level

Example.

X : individual.



p-level

x : pop. dens.

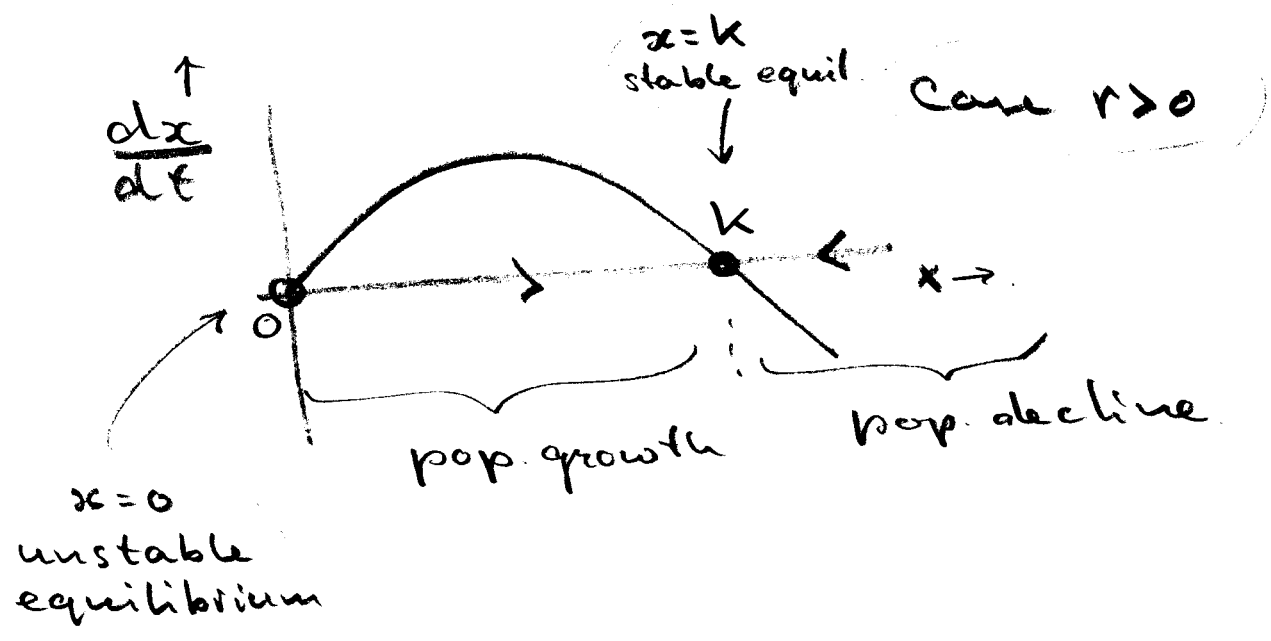
$$\frac{dx}{dt} = \underbrace{-\lambda x}_{\substack{\text{left} \\ \text{(reactant)}}} + \underbrace{2\lambda x}_{\substack{\text{right} \\ \text{(product)}}} - \underbrace{\mu x}_{\substack{\text{left} \\ \text{(reactant)}}} - \underbrace{\nu x^2}_{\substack{\text{left} \\ \text{(reactant)}}} + \underbrace{\frac{1}{2}\nu x^2}_{\substack{\text{right} \\ \text{(product)}}}$$

$$= (\lambda - \mu)x - \frac{1}{2}\nu x^2$$

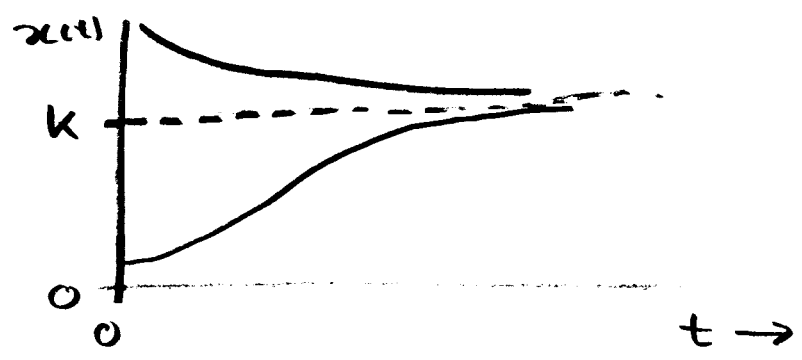
model analysis

Rewrite as

$$\left\{ \begin{aligned} \frac{dx}{dt} &= r x \left(1 - \frac{x}{k}\right) \quad (\text{logistic}) \\ r &= \lambda - \mu \\ k &= \frac{2(\lambda - \mu)}{\nu} \end{aligned} \right.$$



⇒ Qualitative behavior of $x(t)$ for two different starting points



Another example

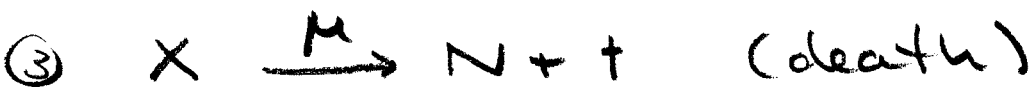
i-level

these are called the i-states.

N : unoccupied territory

X : occupied territory
(= territory owner)

Y : individual without territory



p-level

$$\left\{ \begin{array}{l}
 \frac{dx}{dt} = \beta ny - \mu x \\
 \frac{dy}{dt} = -\beta ny + \alpha x - \nu y \\
 \frac{dn}{dt} = -\beta ny + \mu x + \nu y
 \end{array} \right.$$

①
②
③
④

analysis

Notice that $\frac{d(n+x)}{dt} = 0$,

$\Rightarrow n+x =: n_0$ is constant

$\left\{ \begin{array}{l} \text{total density of territories,} \\ \text{occupied or unoccupied} \end{array} \right\}$.

Use this to eliminate n from the equations:

$$\Rightarrow \begin{cases} \frac{dx}{dt} = \beta(n_0 - x)y - \mu x \\ \frac{dy}{dt} = -\beta(n_0 - x)y + \alpha x - \gamma y \end{cases}$$

Next lecture we analyze this system further...